

Presentation Overview

- CO₂ capture introduction
- Post-combustion capture: State of the art
- CASTOR Integrated Project
- Post-combustion capture: R&D trends
- **q** Final comments



Zagreb, 27 February 2007

CO₂ capture introduction

What are challenges for CO₂ capture?

Capture of CO₂ can be done with technologies presently available but:

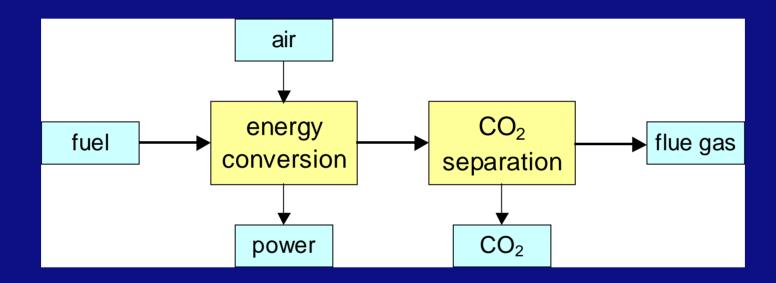
- § Sizeable efficiency reduction
- § Power generation costs will increase
- § There is no experience with CO₂ capture at the power plant scale

Therefore the following questions need to be addressed:

- § How to reduce the additional power consumption as a result of the capture process?
- § How to reduce the costs of the capture?
- § How to make these processes reliable when integrated with power generation?



Post-combustion CO₂ capture CO₂-removal from flue gases



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Why post-combustion capture?

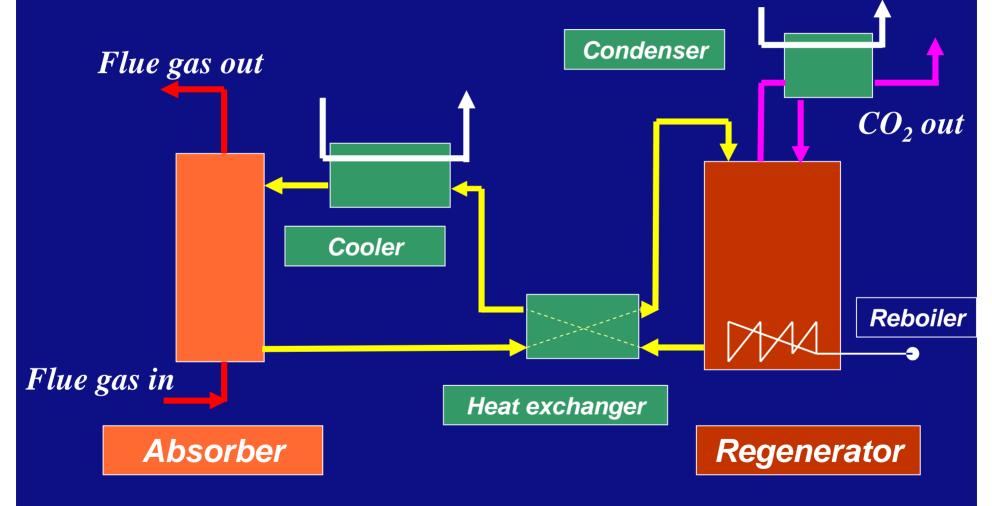
- Add-on to existing power plants and plant concepts
- Capture technologies available, i.e. solvent technologies, which are proven on a smaller scale
- Similarities with cogeneration plant lead to proven methods for integration
- Capture readiness easy to incorporate into power plant tackling issue with infrastructure inertia
- Flexibility in switching between capture no capture
- Capture process development
- Carning by searching will lead to better solvents and solvent processes



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Post-combustion capture: State of the art

Solvent process flow sheet Aqueous ethanolamine solutions





Ethanolamine reaction chemistry: General

Primary, secondary amines (MEA, DEA)

Primary: $CO_2 + 2RNH_2 \hat{a} RNH_3^+ + RNHCOO^-$ Secondary: $CO_2 + 2RNH \hat{a} RNH_2^+ + RNCOO^-$

- A carbamate with varying degrees of stability is formed
- High mass transfer
- Max. loading: 0.5 mol/mol amine
- Tertiary amines, sterically hindered amines (TEA, MDEA, AMP)

Tertiary:
$$CO_2 + RN + H_20 \ \text{a} \ RNH^+ + HCO_3^-$$

Primary/sterically hindered: $CO_2 + RNH_2 + H_20 \ \text{a} \ RNH_3^+ + HCO_3^-$

- Formation of bicarbonate in presence of water
- Low mass transfer
- Max. loading: 1.0 mol/mol amine

State of the art post-combustion CO₂-capture

- **G** Fluor Daniel Econamine FGSM
 - § 30% MEA solution incorporating additives to control corrosion
 - § > 20 commercial plants in sizes between 0.2 and 15 tonnes CO₂/h
- **q** ABB-Lummus
 - § 15-20% MEA solution
 - § 4 commercial plants in size between 6 and 32 tonnes CO_2/h
- Mitsubishi Heavy Industries
 - § KS-1– sterically hindered amines
 - § 2 commercial plants: 9 tonne CO₂/h
- More to come.....



Issues for post-combustion CO₂-capture

- Solvent technologies are leading option but currently:
 - § Power cost increases >50%
 - § Generation efficiency decreases by 15 − 35%
- Solvent process break-throughs required
 - § Energy requirements
 - § Reaction rates
 - § Contactor improvements
 - § Liquid capacities
 - § Chemical stability/corrosion
 - § Desorption process improvements
 - § Hence cost reductions
- Integration with power plant
 - § Heat integration with other process plant, particularly in relation to desorption process
 - § Power plant concepts prepared for incorporation of capture plant





CASTOR

CO₂, from Capture to Storage

a European Integrated Project (IFP – Project leader)



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Consortium and finance

R&D Oil & Gas STATOIL (NO) IFP (FR) TNO (NL) GDF (FR) SINTEF (NO) REPSOL (SP) **ENITecnologie (IT)** NTNU (NO) ROHOEL (AT) BGS (UK) BGR (DE) **BRGM (FR) GEUS (DK) IMPERIAL (UK)** OGS (IT)

Power Companies
VATTENFALL (SE)
ELSAM (DK)
ENERGI E2 (DK)
RWE (DE)
PPC (GR)
E.ON-UK (UK)

Manufacturers
ALSTOM POWER (FR)
MITSUI BABCOCK (UK)
SIEMENS (DE)
BASF (DE)
GVS (IT)

Sponsor: ELECTRABEL (BE)

Budget: 15.8 M€ EU funding: 8.5 M€

STUTTGARTT U. (DE)

TWENTE U. (NL)

Industrial funding: 2.2 M€

Duration: 4 years

Co-ordinator: IFP

Chair of the Executive Board: Statoil

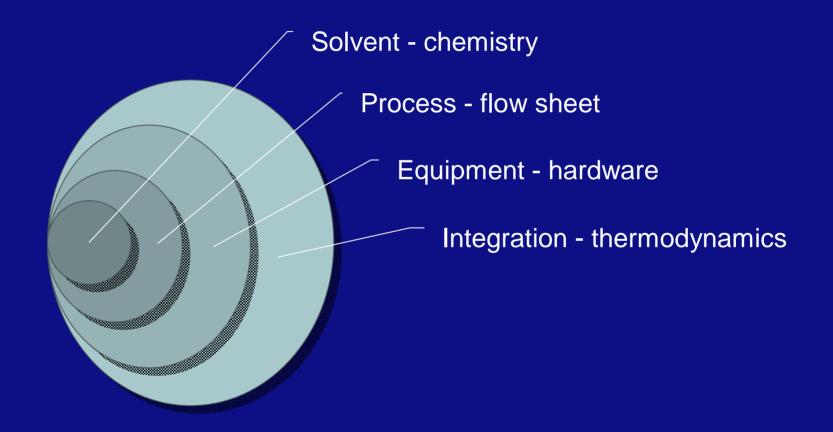


CASTOR Objectives / targets

- Reduce the cost of CO₂ post-combustion capture
- Contribute to the feasibility & acceptance of the geological storage concept
- Validate the concept on real site(s)
 - Pilot testing for capture (25 t CO₂ / day)
 - Follow-up of ongoing storage projects

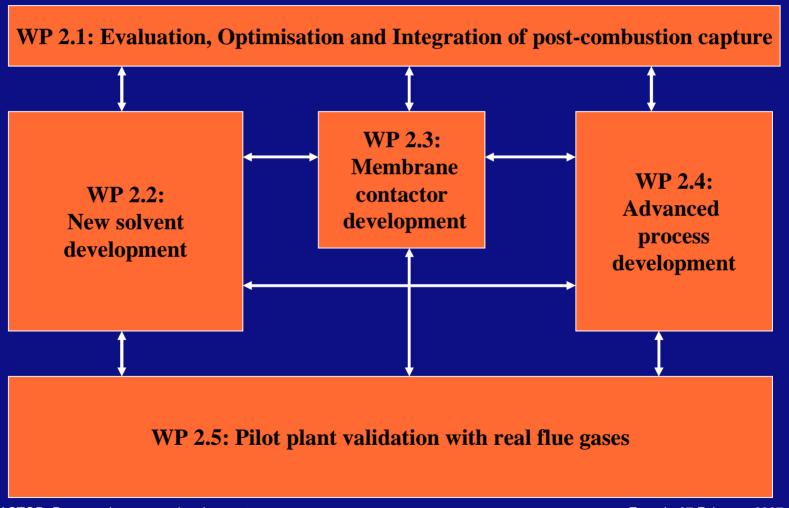


Integrated approach in solvent process development for post-combustion capture



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SP2 – Work package structure



Solvent selection criteria

- Cyclic capacity
- Regeneration energy requirement
- Rate of reaction/Mass transfer
- Molecular transport properties
- Vapor pressure
- Corrosiveness
- Foaming properties
- Chemical stability
- Toxicity
- Water solubility
- Cost and availability



Solvent development procedure



8 solvents selected

30 solvents pre-selected

Pilot plant experiments

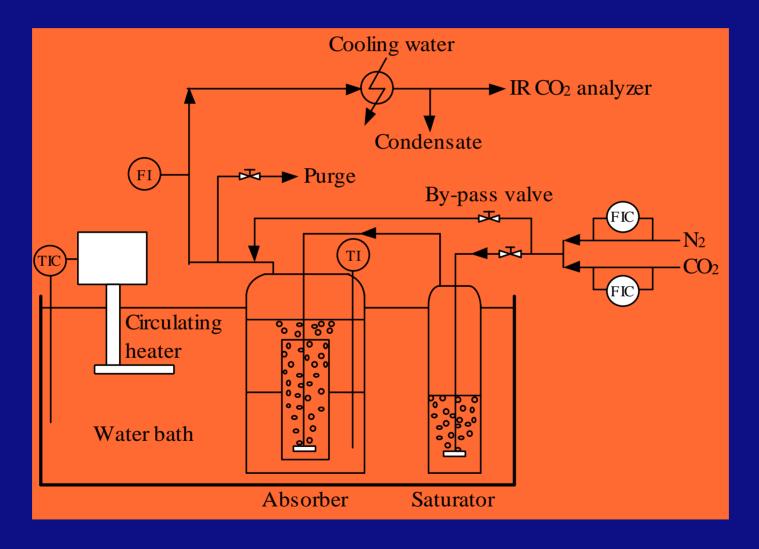
Lab-pilot experiments

Process design studies
Degradation studies
Corrosion studies
Solvent characterisation

Solvent screening studies

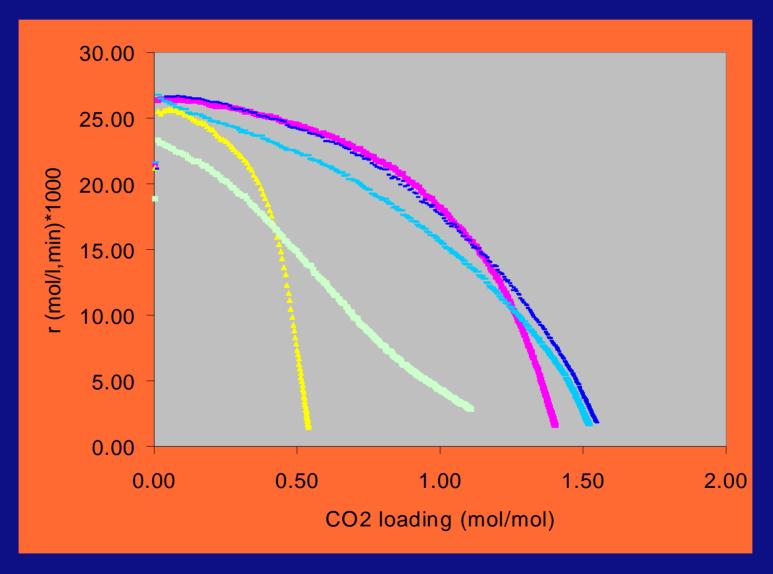


Screening apparatus for CO₂ absorption





Screening results - triamines





Experimental set-ups for solvent selection



Miniplant Stuttgart Uni.







Equilibrium
Apparatus
SINTEF/NTNU

Corrosion test cell IFP





Overview of consequences for European power plants CASTOR Results for base cases (30% MEA)

Item	Increase in investment	CoE Increase	Efficiency decrease	Cost per tonne CO ₂ avoided
Gas fired combined cycle [393 MW _e]	48 %	51 %	18 %	54 €
Bituminous coal fired power plant [600 MW _e]	25 %	84 %	31 %	40 €
Lignite fired power plant [1000 MW _e]	17 %	88 %	34 %	40 €



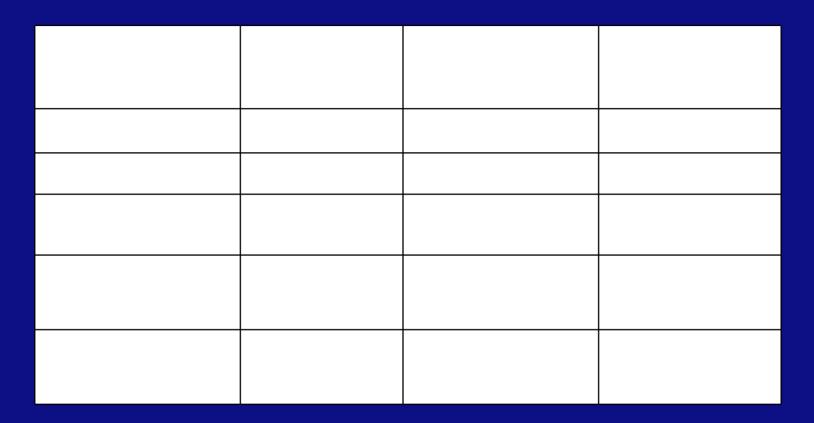
Contributions to increase in CoE CASTOR base case results

- CoE increases due to decreased power output
- CoE increases due additional investments

Power plant	CoE increase due to capital costs	CoE increase due to efficiency loss
GTCC	60 %	40 %
Solid fuel power plants	40 %	60 %



Clean-up of flue gases - SO₂ capture



- Wet processes (liquor/slurry) achieve highest removal
- Technologies are available to achieve this
- Economics of these technologies have been mapped



FGD: Life cycle costs and marginal capture costs

	Bituminous 600 MW	Lignite 1000 MW	Lignite 380 MW
Limit	Cost Increase € tonne CO ₂	Cost Increase ∉ tonne	Cost Increase € tonne CO ₂
200 mg/Nm ³	0.000	0.000	0.000
50 mg/Nm ³	0.053	0.079	0.130
10 mg/Nm ³	0.077	0.151	0.231

EU Post-combustion capture test facility in Esbjerg Denmark (Dong Energy)

- O Characteristics
 - § Scale 1 tonne/h CO₂; 5000 m³/h flue gas
 - § Flue gases from coal firing
- Experimental programme
 - Solvent process validation for existing solvents
 - Solvent process validation for new solvents







Energy impacts for post-combustion CO₂ capture using solvent processes

- Thermal energy
 - § Regeneration of solvents; Extracted from steam cycle in power plant:
 - Sum of solvent heating, desorption enthalpy and reflux ratio
 - Energy impact determined by solvent properties, process design and integration into power plant
- Electricity
 - § Flue gas fans:
 - Higher CO₂-content in flue gas reduces specific power consumption
 - § Solvent pumps, cooling water pumps:
 - Higher CO₂-loading of solvent reduces specific power consumption
 - § CO₂ compressor:
 - Specific power consumption determined by pressure ratio



Post-combustion capture process performances: Past, present and future

Year	1995	2005	2015
Thermal energy	4.2 GJ/tonne CO ₂	3.2 GJ/tonne CO ₂	2.0 GJ/tonne CO ₂
Power equivalent (Factor used)	0.292 kWh/kg CO ₂ (0.25)	0.178 kWh/kg CO ₂ (0.20)	0.083 kWh/kg CO ₂ (0.15)
Power for capture	0.040 kWh/kg CO ₂	0.020 kWh/kg CO ₂	0.010 kWh/kg CO ₂
CO ₂ compressor	0.114 kWh/kg CO ₂	0.108 kWh/kg CO ₂	0.103 kWh/kg CO ₂
Total	0.446 kWh/kg CO ₂	0.306 kWh/kg CO ₂	0.196 kWh/kg CO ₂

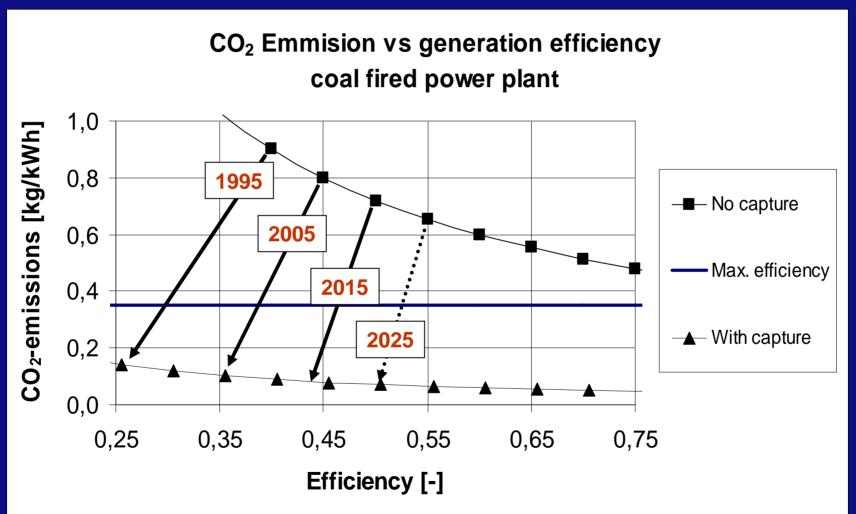


Post-combustion capture in a coal fired power plant (Emission factor: 0.1 tonne CO₂/GJ)

Year	1995	2005	2015
Base plant efficiency	40 %	45 %	50 %
Base plant emission	900 kg CO ₂ /MWh	800 kg CO ₂ /MWh	720 kg CO ₂ /MWh
Power loss due to capture	0.446 kWh/kg CO ₂	0.306 kWh/kg CO ₂	0.196 kWh/kg CO ₂
Plant efficiency with 90% capture	25.5 %	35.1 %	43.6 %
Emission with 90% capture	141 kg CO ₂ /MWh	103 kg CO ₂ /MWh	82 kg CO ₂ /MWh



Development of capture technology hand in hand with generation efficiency improvements





CAPRICE (Awarded in last EU FP6 call)

- <u>CO</u>₂ capture using <u>A</u>mine <u>Pr</u>ocesses: <u>I</u>nternational <u>C</u>ooperation and <u>E</u>xchange
- Cooperation between CASTOR core partnership with consortium around International Test Centre on CO₂ capture (University of Regina, Canada)
- TNO, NTNU, Stuttgart University, IFP, Elsam, Energie E2, E.ON-UK
- Other CSLF countries involved: China, Russia, Brazil
- Budget: 1.144 M€ (0.383 EC contribution)
- Period: 2 years

Project activities:

- 1. Benchmarking and validation of amine processes
- 2. Membrane contactor validation studies
- 3. Development of tools for integration



Post-combustion capture: Research and development trends

Overview of Research and Development trends

- Formulated amines
 - Mixtures of "promoters and loaders"
 - Corrosion inhibitors added
- Multiple amine groups
 - Diamines, triamines
 - Hyperbranched polymers
- Non-aqueous solvents
 - Ionic liquids
- Modified process concepts
 - Intercooling
 - Heat exchange integration in stripper
 - Integration of compression
 - Split flow
- Novel process components
 - Membrane contactors for absorption/desorption
- Robust solvents
 - Ammonia



Final Comments

- Post-combustion capture presents the decarbonisation approach with least impact on power generation processes
- Relatively simple to take incorporate capture plant in power plant for post-combustion capture
- Capture by "doing" and "searching"
- R&D programmes require an integrated approach to postcombustion capture
- Several advanced processes under development show opportunities for efficiency improvement and cost reduction

